Density

Which is heavier, a ton of feathers or a ton of bricks?

Overview of Density

- Describes the degree of compactness of a substance.
- Density is mass per unit volume

$$=\frac{m}{V}$$

- Common Units: g/cc, kg/liter, kg/m³
- Related to sinking and floating.

Density & Superposition

- Superposition: More dense substances will sink beneath less dense substances.
- An example of this is Oil and Water.
 - The water sinks because it is more dense than the oil
- Misconception!
 - Water is heavier than oil.



©1998 Science Joy Wagon

Objects that change density

- Lava Lamp volume
 - hot light expands waxy goo
- Submarines mass
 - pump water in or out
- Blimps & Hot Air Balloons mass
 - maintain volume but let air in or out
- People Swimming volume
 - expand or collapse lungs
- Stoned Crocodiles? mass

4

Misconceptions, Exceptions & Common Mistakes

- Misconception
 - Just because something is more dense doesn't mean it is heavier.
- Exception
 - Solid objects don't usually undergo superposition.
 - Example: Car on Asphault
- · Common Mistakes
 - When using the formula to solve for Volume care must be taken with algebra

©1998 Science Joy Wagon